In the Claims:

This Listing of Claims Will Replace All Prior Versions, And Listings, Of Claims In The Application

1. (Original) A process for the preparation of a compound of formula (I)

$$R^{6} \xrightarrow{X} R^{3} \qquad (I)$$

wherein

X is selected from CH_{2 Or O};

R¹ is selected from the group consisting of hydrogen and C₁₋₄alkyl;

R³, R⁴, R⁵ and R⁶ are each independently selected from hydrogen or lower alkyl and, when X is CH₂, R⁵ and R⁶ may be alkene groups joined to form a benzene ring and, when X is O, R³ and R⁴ and/or R⁵ and R⁶ together may be a methylenedioxy group of the formula:

wherein

R⁷ and R⁸ are same or different and are hydrogen, lower alkyl or are alkyl and are joined to form a cyclopentyl or cyclohexyl ring;

comprising

$$R^{6} \xrightarrow{X} R^{3} \qquad H_{2}N \xrightarrow{S} NH_{2} \qquad R^{6} \xrightarrow{X} R^{3} \qquad (II)$$

reacting a compound of formula (II) with sulfuryl diamide, at an elevated temperature, in the presence of from 0 to about 10% water, to yield the corresponding compound of formula (I).

- 2. (Original) The process of Claim 1, wherein the compound of formula (II) is diacetone fructose.
- 3. (Original) The process of Claim 2, wherein the sulfuryl diamide is present in an amount greater than about 0.9 equivalent
- 4. (Original) The process of Claim 3, wherein the sulfuryl diamide is present in an amount equal to about 1.5 to about 3 equivalents.
- 5. (Original) The process of Claim 2, wherein the compound of formula (II) is reacted with sulfuryl diamide in the presence of a non-aqueous organic or inorganic base.
- 6. (Original) The process of Claim 5, wherein the non-aqueous organic or inorganic base is a tertiary amine base.
- 7. (Original) The process as in Claim 6, wherein the tertiary amine base is pyridine.
- 8. (Original) The process of Claim 5, wherein the non-aqueous organic or inorganic base is present in an amount greater than about 1 equivalent.
- 9. (Original) The process of Claim 8, wherein the non-aqueous organic or inorganic base is present in an amount equal to about 3 to about 5 equivalents.
- 10. (Original) The process of Claim 2, wherein the compound of formula (II) is reacted with sulfuryl diamide in an aprotic organic solvent.
- 11. (Original) The process of Claim 10, wherein the aprotic organic solvent is a non-aqueous organic base.

- 12. (Original) The process of Claim 11, wherein the non-aqueous organic base is pyridine.
- 13. (Original) The process of Claim 2, wherein the elevated temperature is in the range of from about 90°C to about 170°C.
- 14. (Original) The process of Claim 13, wherein the elevated temperature is in the range of from about 120°C to about 140°C.
- 15. (Original) The process of Claim 2, wherein the compound of formula (II) is reacted with sulfuryl diamide, in the presence of from 0 to about 3% water.
- 16. (Original) A compound prepared according to the process of Claim 1.
- 17. (Original) A process for the preparation of a compound of formula (la)

comprising

reacting a compound of formula (IIa) with sulfuryl diamide, at an elevated temperature, in the presence of from 0 to about 10% water, to yield the corresponding compound of formula (Ia).

- 18. (Original) The process of Claim 17, wherein the sulfuryl diamide is present in an amount greater than about 0.9 equivalents.
- 19. (Original) The process of Claim 18, wherein the sulfuryl diamide is present in an amount equal to about 1.5 to about 3 equivalents.
- 20. (Original) The process of Claim 17, wherein the compound of formula (IIa) is reacted with sulfuryl diamide in the presence of an non-aqueous organic or inorganic base.
- 21. (Original) The process of Claim 20, wherein the non-aqueous organic or inorganic base is a tertiary amine base.
- 22. (Original) The process as in Claim 21, wherein the tertiary amine base is pyridine.
- 23. (Original) The process of Claim 20, wherein the non-aqueous organic or inorganic base is present in an amount greater than about 1 equivalent.
- 24. (Original) The process of Claim 23, wherein the non-aqueous organic or inorganic base is present in an amount equal to about 3 to about 5 equivalents.
- 25. (Currently Amended) The process of Claim 17, wherein the the compound of formula (IIa) is reacted with sulfuryl diamide in an aprotic organic solvent.

- 26. (Original) The process of Claim 25, wherein the aprotic organic solvent is a non-aqueous organic base.
- 27. (Original) The process of Claim 26, wherein the non-aqueous organic base is pyridine.
- 28. (Original) The process of Claim 17, wherein the elevated temperature is in the range of from about 90°C to about 170°C.
- 29. (Original) The process of Claim 28, wherein the elevated temperature is in the range of from about 120°C to about 140°C.
- 30. (Original) The process of Claim 17, wherein the compound of formula (IIa) is reacted with sulfuryl diamide in the presence of from 0 to about 3% water.
- 31. (Original) A compound prepared according to the process of Claim 17.
- 32. (Original) A pharmaceutical composition comprising a pharmaceutically acceptable carrier and the compound according to Claim 16.
- 33. (Original) A pharmaceutical composition made by mixing a pharmaceutically acceptable carrier and the compound according to Claim 16.
- 34. (Original) A process for making a pharmaceutical composition comprising mixing a pharmaceutically acceptable carrier and the compound according to Claim 16.
- 35. (Original) A pharmaceutical composition comprising a pharmaceutically acceptable carrier and the compound according to Claim 31.

- 36. (Original) A pharmaceutical composition made by mixing a pharmaceutically acceptable carrier and the compound according to Claim 31.
- 37. (Original) A process for making a pharmaceutical composition comprising mixing a pharmaceutically acceptable carrier and the compound according to Claim 31.